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Selective C-2 and C-4 Deacylation and Acylation of Taxol: The First Synthesis of a C-4 Substituted Taxol Analogue

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Abstract: Hydrolytic procedures for selective 2-debenzoylation and 2,4-dideacylation of 2'-*O*-*tert*-butyldimethylsilyl-7-*O*-(triethylsilyl)taxol are reported. The first synthesis and biological evaluation of a 4-substituted analogue, 4-deacetyl-4-isobutanoyltaxol, is presented. The chemistry described in this letter is suitable for the facile synthesis of taxol congeners modified at C-2 and/or C-4.

The clinical efficacy,¹ chemistry² and structure-activity relationships^{3,4} of the anticancer drug taxol (**1**) isolated from the western yew, *Taxus brevifolia*,⁵ have been studied extensively in recent years. Most structure-activity relationships studies have focused on the synthesis and biology of C-13 phenylisoserine side chain analogues.^{3,4,6} However, more recently a number of reports have appeared, detailing the effects of structural alterations on the diterpene moiety of taxol on biological activity.^{3,4,6} These studies revealed that modification at the southern part of the taxol molecule (C-1 to C-5) typically effect the biological activity more than alterations at the northern part (C-6 to C-12).^{3,4,6}

In this letter we are reporting synthetic strategy for the preparation of 2-modified analogues which is superior to a recently reported method⁷ for the selective 2-debenzoylation of silyl protected taxol.^{8,9} In addition, we are detailing the synthesis and biological activity of the first 4-substituted taxol analogue, 4-deacetyl-4-isobutanoyltaxol (**11**).

In a continuation of our studies on the selective hydrolysis of baccatin III derivatives,⁹ we have found that 2'-*O*-*tert*-butyldimethylsilyl-7-*O*-(triethylsilyl)taxol (**2a**) can be selectively hydrolysed with anhydrous KOH¹⁰ to yield the corresponding 2-debenzoyl- and 2,4-dideacyltaxol derivatives **3a** and **6**, respectively (Scheme). When analogue **2a** was treated with 1.2 equiv of anhydrous KOH at -15 °C for 24 h, 2-debenzoyltaxol **3a** was obtained in 70-80% yield (Scheme). Utilization of 4 equivalents of anhydrous KOH at -20 °C for 72 h (Scheme) gave 75% of 4-deacetyl-2-debenzoyl-2'-*O*-*tert*-butyldimethylsilyl-7-*O*-(triethylsilyl)taxol (**6**).

The starting material (**2a**) for the synthesis of these derivatives is either generated by silyl protection of taxol (Scheme) or semisynthetically by reacting 7-*O*-(triethylsilyl)baccatin III with the appropriate optically active *N*-benzoyl β -lactam.¹¹ The silyl protecting group at the 7-hydroxyl group in **2a** is needed to avoid the well known base induced epimerization at C-7.¹² The rationale for our choice of protecting group at 2' is the generation of steric hindrance at the C-13 ester group, thereby avoiding its hydrolysis. Previous studies on taxol and baccatin III¹³ had demonstrated that mild methanolysis results in the hydrolysis of the esters at C-10 and C-13.^{5,14} The choice of placing a *tert*-butyldimethylsilyl protecting group at the 2' hydroxyl turned out to be of

critical importance for the success of our reactions. For example, submitting 2',7-*O*-bis(triethylsilyl)taxol (**2e**) to the same reaction conditions led to a mixture of hydrolysis products, including C-13 side chain cleaved derivatives.⁷ Our earlier studies⁹ with baccatin III derivatives had demonstrated that the 2-benzoate group could be selectively hydrolysed with *tert*-BuOK while retaining the esters at C-10 and C-4. We believe that the selective 2-benzoate hydrolysis of taxol and baccatin III derivatives is facilitated by assistance of the neighboring C-1 hydroxyl. Debenzoylation of **2a** was also achieved with *tert*-BuOK at -15 °C for 72 h. The addition of H₂O (formation of anhydrous KOH) enhanced the reaction rate (24 h instead of 72 h). Yields and product distribution were similar under both reaction conditions. In addition to 2-debenzoyl derivative **3a** (70-80% yield), we also isolated about 15% 2,4-dideacyl derivative **6**. The 4-acetate seems to be cleaved only in the absence of the 2-benzoate, suggesting assistance in its hydrolysis from the 2-hydroxyl. It is of interest to note that the analogous 2-debenzoylation of 10-acetyl-2'-*O*-(*tert*-butyldimethylsilyl)taxotere (**2b**) gave 85-89% yield of **3b**. No formation of 2,4-dideacylated product was observed. Standard acylation of **3a** with DCC/DMAP and 4-chlorobenzoic acid in toluene at 55 °C, followed by silyl deprotection gave 2-(4-chlorobenzoyl)-2-debenzoyltaxol (**5**) in 65% overall yield.

Since 2,4-dideacyl derivative **6** was observed as a byproduct in the 2-debenzoylation reaction, we explored whether **6** could be obtained as the major product by a change of reaction conditions. We found that using 2-3 equivalents of anhydrous KOH at -20 °C for 72 h in the hydrolysis reaction gave 75% of 4-deacetyl-2-debenzoyl-2'-*O*-(*tert*-butyldimethylsilyl)-7-*O*-(triethylsilyl)taxol (**6**).

We then investigated the conversion of derivative **6** to 4-acetyl modified taxol analogue **11** (Scheme). As precedented in recent taxol total syntheses,^{15,16} the introduction of the acetyl group at the sterically encumbered 4-hydroxy group of baccatin III derivatives is possible after formation of a 1,2 cyclic carbonate. Thus, reaction of **6** with 1,1'-carbonyldiimidazole provided cyclic carbonate **7**, which was acylated at the C-4 hydroxyl group using isobutyric anhydride and DMAP to form acylated derivative **8** in 67% yield. Regioselective opening of the carbonate in **8** with PhLi (formation of **9** and **10**) followed by silyl deprotection led to 4-deacetyl-4-isobutanoyltaxol (**11**) in 46% and 4,10-dideacetyl-4-isobutanoyltaxol (**12**) in 18% yield.¹⁷ It is noteworthy that treatment of **8** with PhLi was relatively selective and did not lead to a significant amount of undesired hydrolysis products. Thus, we have demonstrated for the first time that the conversion of the 1,2-cyclic carbonate with PhLi to its 2-benzoate analogue can be achieved in the presence of the C-13 phenylisoserine side chain.

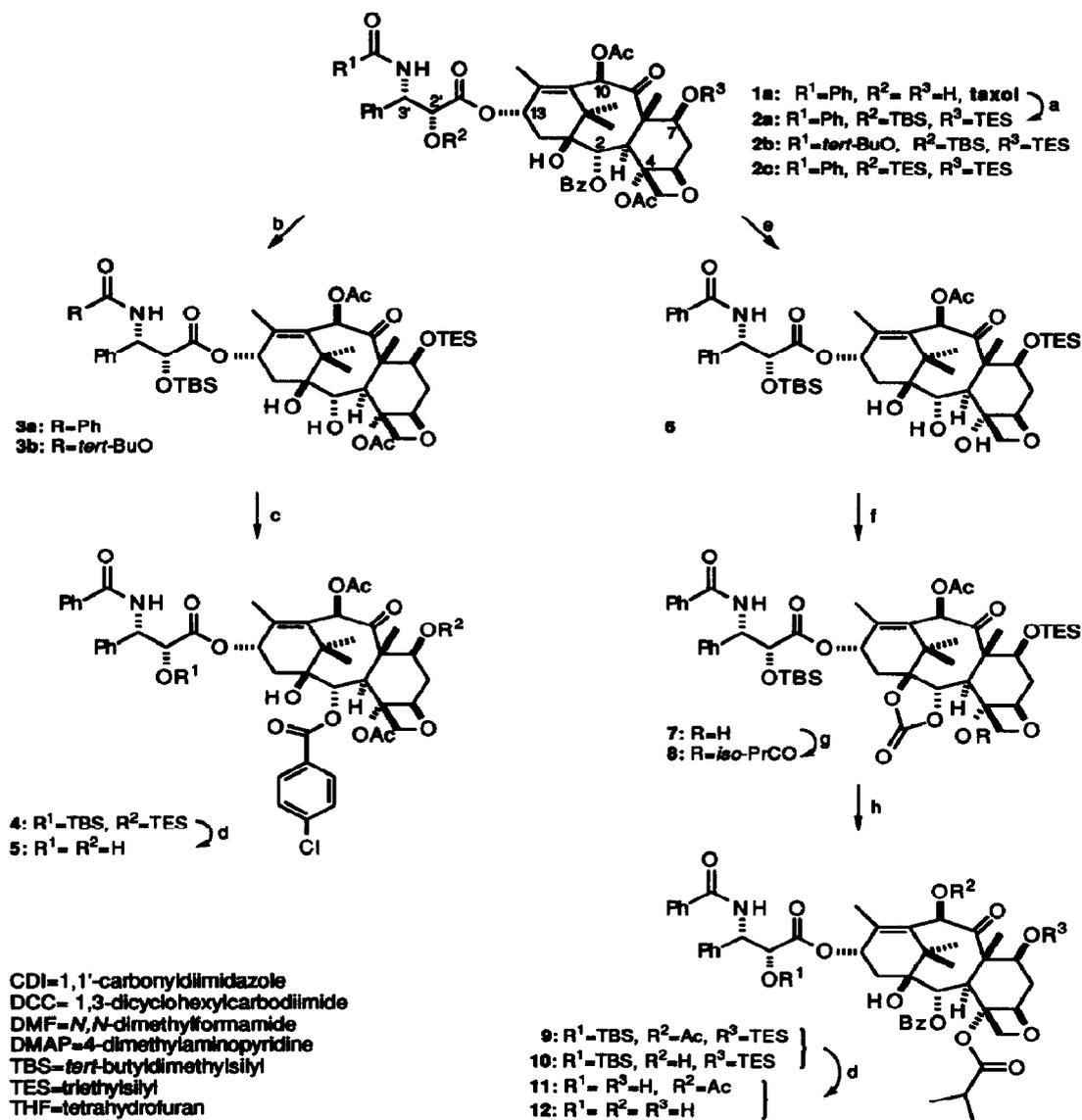
Derivatives **5** and **11** were investigated for their ability to stimulate the assembly of microtubules and for their cytotoxicity against B16 melanoma cells (Table). Derivative **5** was found to be slightly more active

Table. *In vitro* biological evaluation of taxol analogues **5** and **11**.^a

compound	microtubule assembly ^b ED ₅₀ /ED ₅₀ (taxol)	B16 melanoma cytotoxicity ^b ED ₅₀ /ED ₅₀ (taxol)
1	1.0	1.0
5	0.79	3.1
11	2.6	4.5

^aFor experimental procedures see ref. 18. ^bData reported relative to taxol = 1.0.

Scheme



(a) i. Imidazole (15 equiv), TBSCl (15 equiv), DMF, rt, 12 h, quant. ii. TESCl (10 equiv), pyridine, rt, 12 h: **2a**, 90%. (b) *tert*-BuOK (1.4 equiv), H₂O (1.2 equiv), THF, -40 to -15 °C, 24 h: **3a**, 70-80%; **3b**, 85-89%. (c) DMAP (12 equiv), DCC (12 equiv), 4-chlorobenzoic acid (12 equiv), toluene, 55 °C, 6 h. (d) Pyridinium hydrofluoride, pyridine, 0 °C to rt, 4-6 h: **5**, 65% (from **3a**); **11**, 48% (from **8**); **12**, 18% (from **8**). (e) *tert*-BuOK (2.4 equiv), H₂O (2.2 equiv), THF, -40 to -20 °C, 48 h, then *tert*-BuOK (1.2 equiv), H₂O (1.1 equiv), THF, -40 to -20 °C, 24 h: **6a**, 75%. (f) CDI (20 equiv), THF, 55 °C, 6 h: **7**, 87%. (g) Isobutyric anhydride (20 equiv), DMAP (20 equiv), CH₂Cl₂, rt, 48 h: **8**, 67%. (h) PhLi (10 equiv), THF, -78 °C, 10 min.

than taxol in the microtubule assembly assay. However, compound **5** was about three times less cytotoxic against B16 melanoma cells, suggesting that the reduced activity is due to a difference in uptake and/or metabolism.¹⁹ The 4-isobutyric derivative **11** was less active than taxol in both assays (Table).

In conclusion, we have developed efficient methodology for the hydrolysis of taxol derivative **2a** to its 2-debenzoyl and 2,4-dideacyl analogues **3a** and **6**, respectively.²⁰ The availability of these products will allow the investigation of structure-activity relationships of C-2 and C-4 modified taxol analogues. An efficient synthetic pathway was devised, suitable for the convergent synthesis of C-4 analogues of taxol.

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- (20) All newly prepared compounds displayed spectroscopic properties in agreement with their structures. Yields are unoptimized.

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